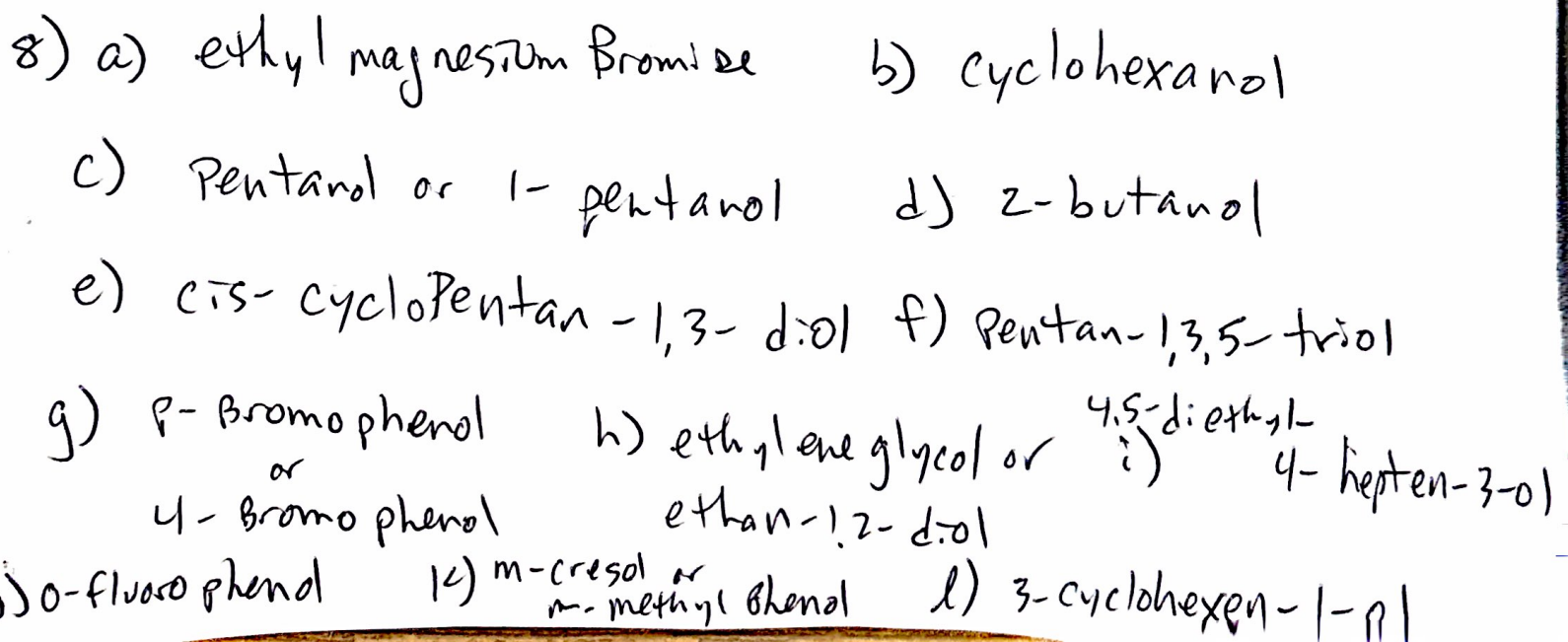
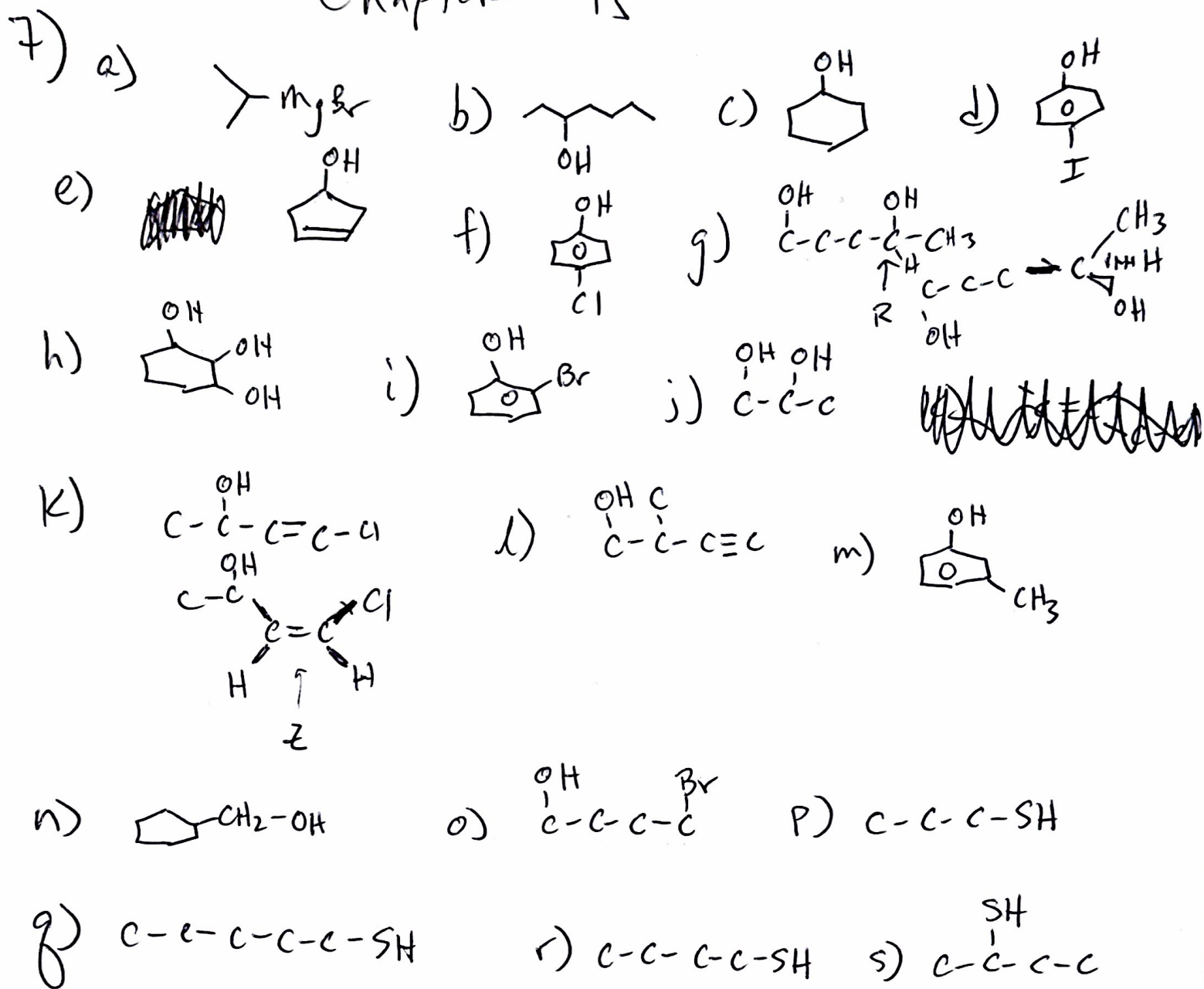
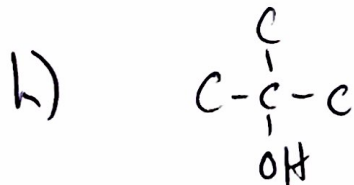
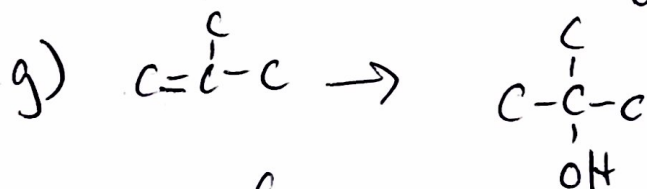
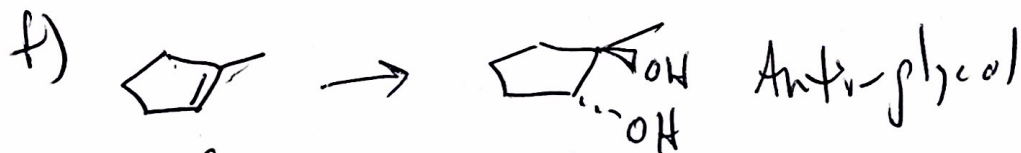
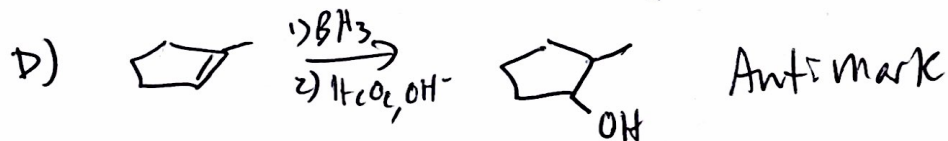
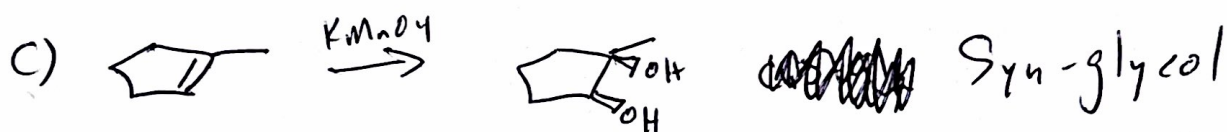
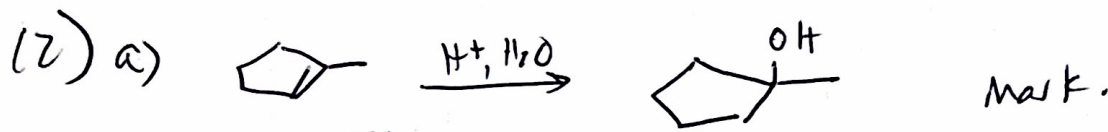
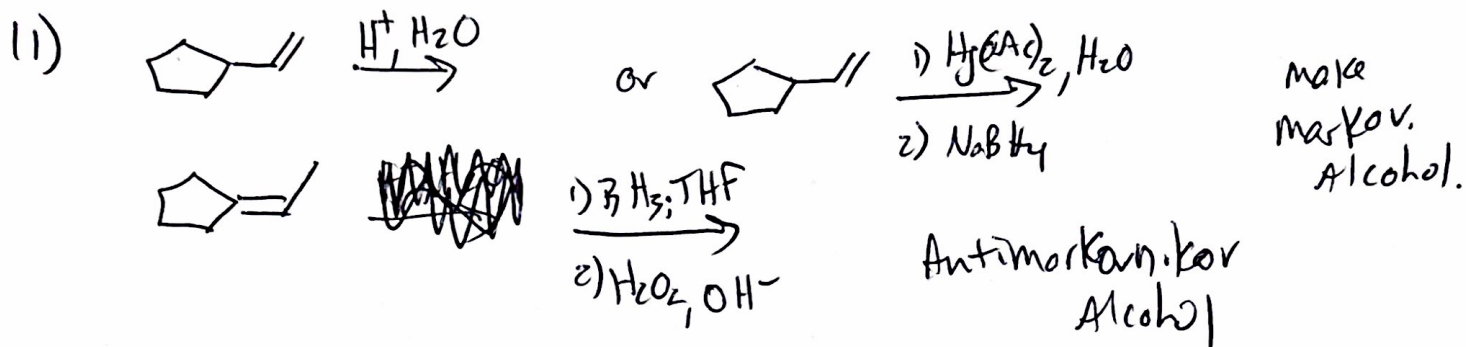


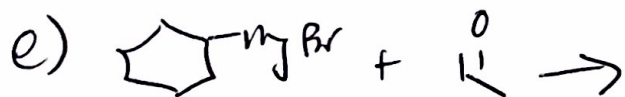
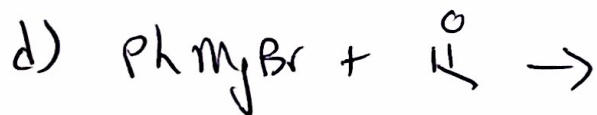
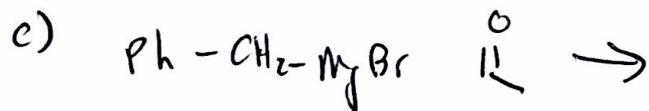
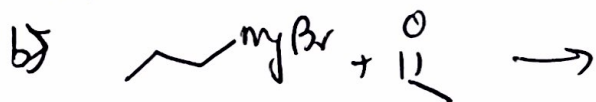
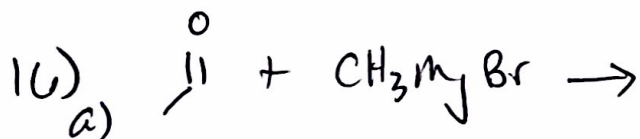
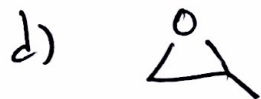
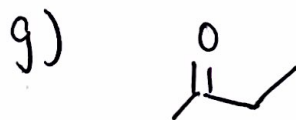
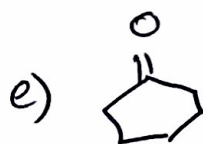
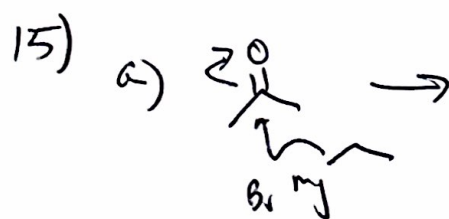
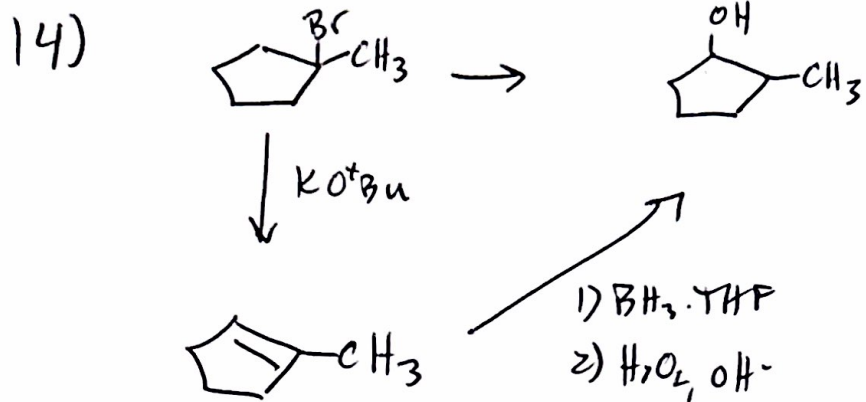
Chapter 13

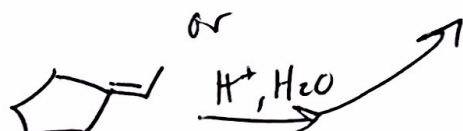
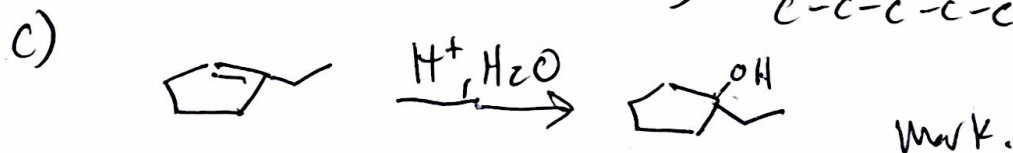
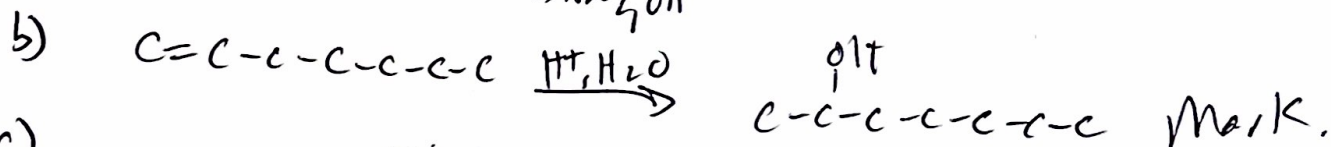
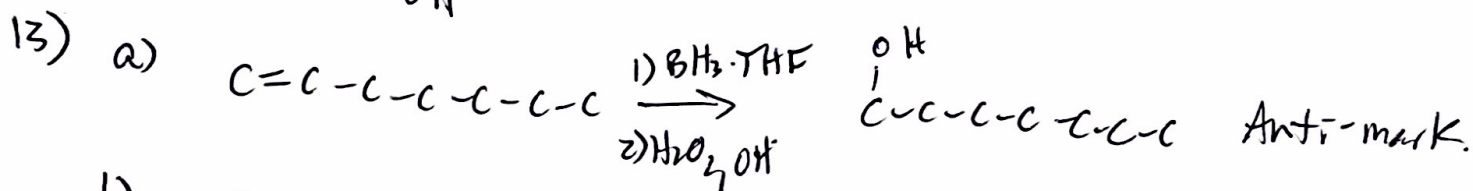
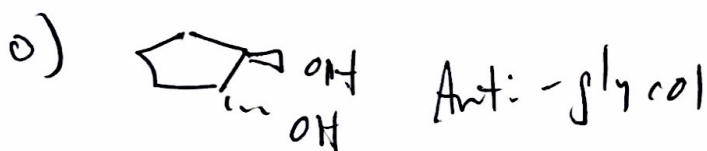
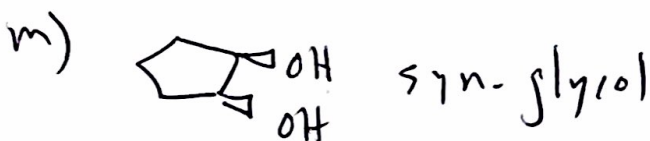
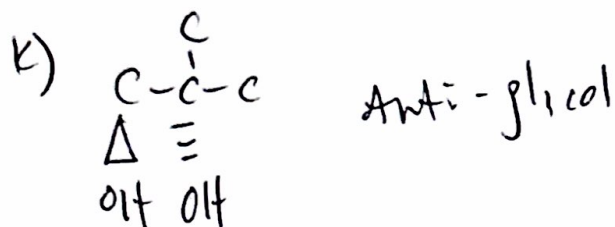
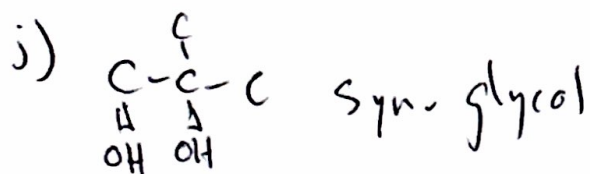
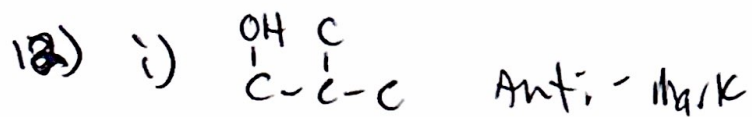


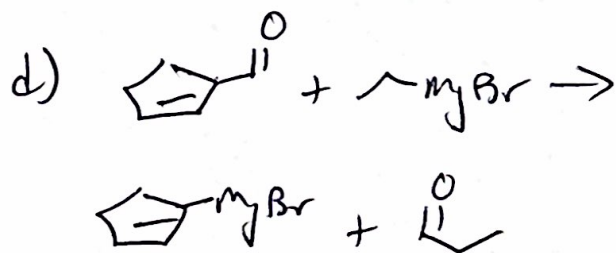
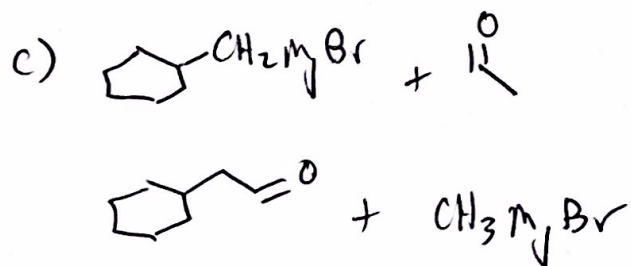
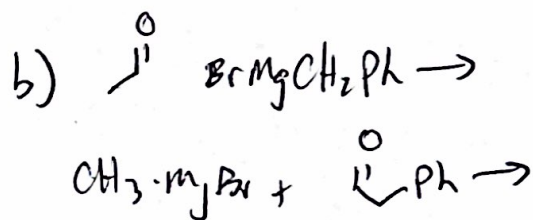
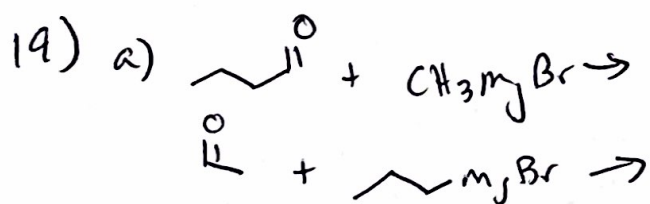
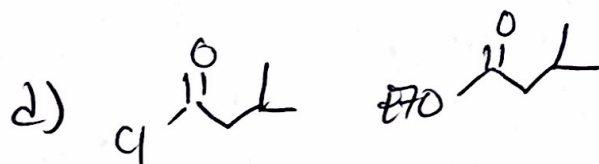
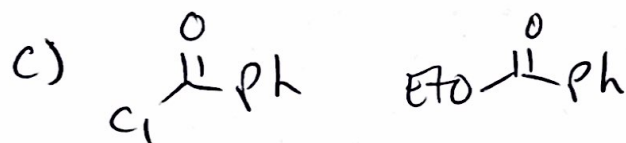
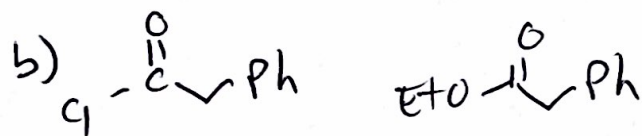
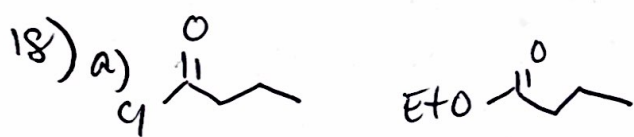
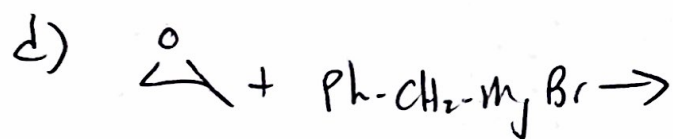
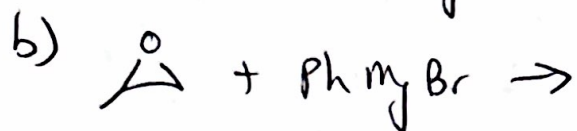
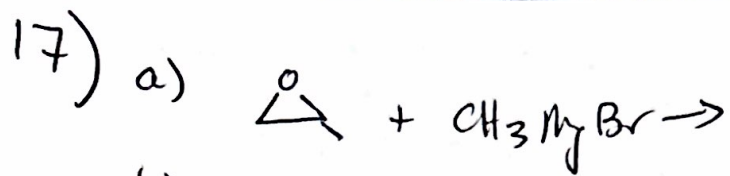
13. 9) orthodox paradox metaphysics parachute
metaphors orthodontist paradise

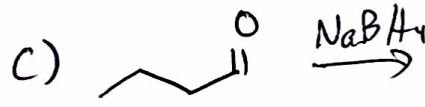
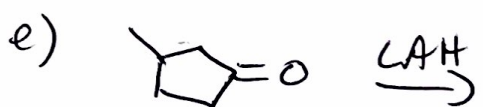
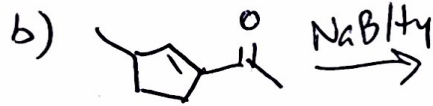
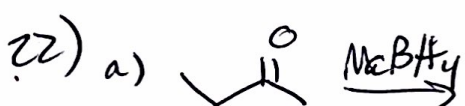
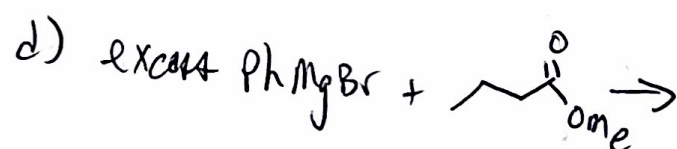
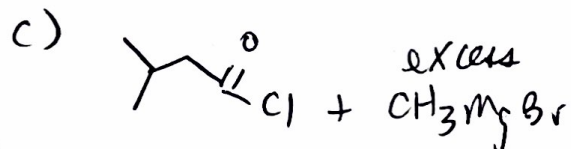
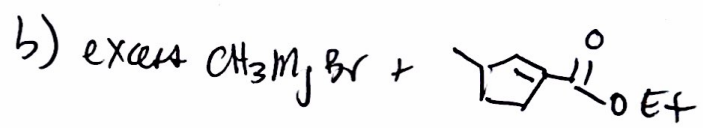
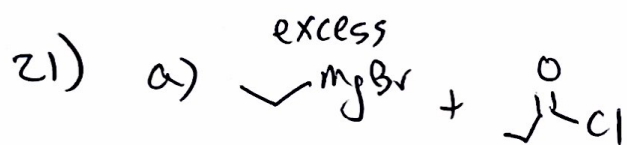
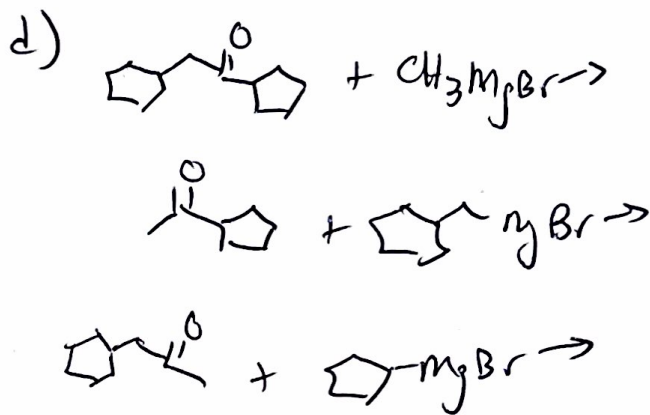
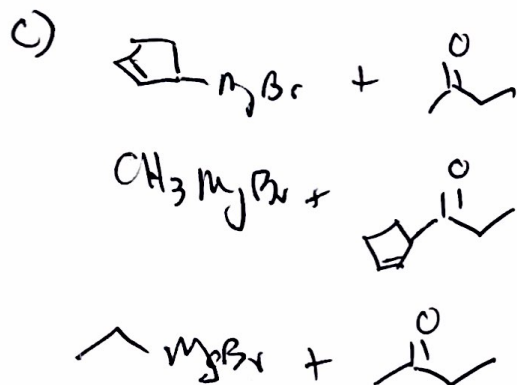
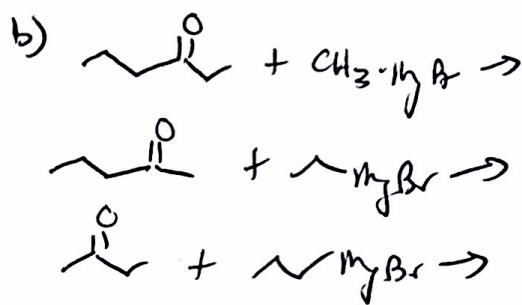
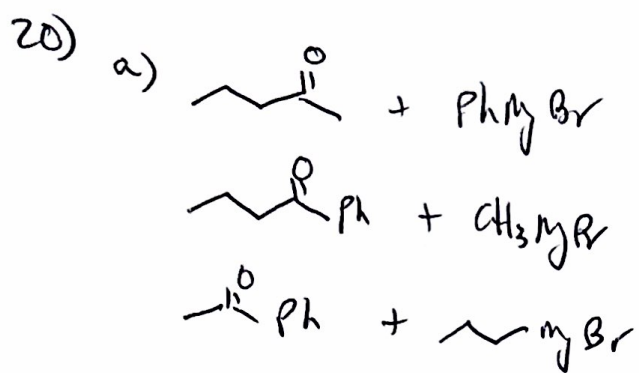
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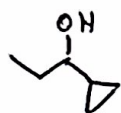
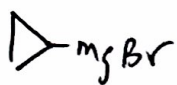








23) a)



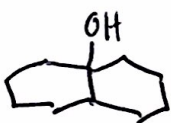
b)

~~CH₄~~ Grignard Reagent
Grabs H on N.

c)

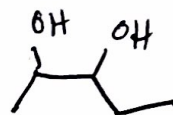


d)



MARK..

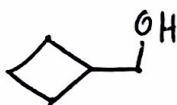
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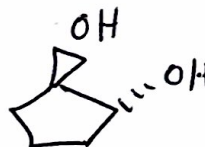
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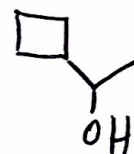
g)



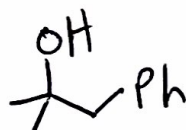
h)



i)



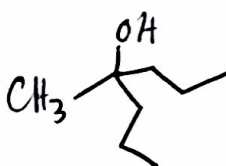
j)



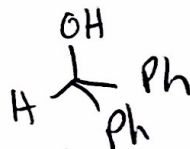
k)



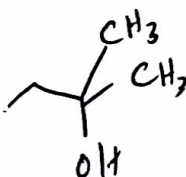
l)



m)



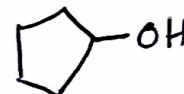
n)



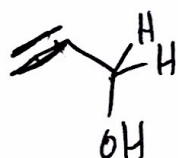
o)



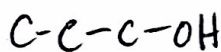
p)



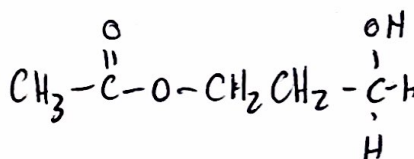
q)



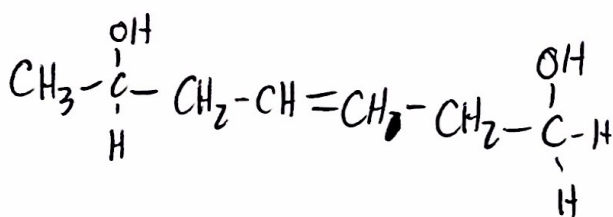
r)



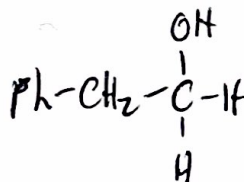
s)



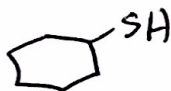
t)



u)



v)



24) a) NaBH_4 2) $\text{H}_2\text{O}, \text{H}^+$ b) 1) LAH 2) $\text{H}^+, \text{H}_2\text{O}$

c) H_2 , Raney Ni d) LAH 2) $\text{H}^+, \text{H}_2\text{O}$

e) NaBH_4 2) H^+

f) H_2 , Raney Ni g) 1) $\text{BH}_3 \cdot \text{THF}$
2) $\text{H}_2\text{O}_2, \text{OH}^-$

h) 1) Li
2) CuI

i) $\text{H}-\text{C}(=\text{O})-\text{Ph}$ j) OsO_4

k) $\text{R}-\text{C}(=\text{O})-\text{O}-\text{O}-\text{H}, \text{H}_2\text{O}$

l) $\text{H}^+, \text{H}_2\text{O}$ or $\text{H}_2\text{O}_2, \text{H}_2\text{O}$
2) NaBH_4

m) $\text{CH}_3\text{CH}_2\text{CH}_2\text{COCl}$ or $\text{CH}_3\text{CH}_2\text{CH}_2\text{COOEt}$

n) $\text{H}-\text{C}(=\text{O})-\text{CH}_2\text{CH}_3$

1) NaNH_2
2) Cyclopropanone

p) 1) NaNH_2
2) $\text{CH}_3\text{CH}_2\text{CHO}$

q) LAH 2) H^+

r) PhMgBr (excess) 2) H^+

s) LAH 2) H^+

t) $\text{H}-\text{C}(=\text{O})-\text{O}-\text{CH}_2\text{CH}_3$ 1) excess PhMgBr
2) H^+ $\rightarrow$ $\text{Ph}-\text{C}(\text{OH})(\text{H})-\text{Ph}$

u) Cyclopropanone 1) CH_3MgBr
2) H^+ $\rightarrow$ Cyclopropanol

v) $\text{Ph}-\text{CH}_2\text{Cl}$ 1) Mg
2) $\text{H}-\text{C}(=\text{O})-\text{H}$
3) H^+ $\rightarrow$ $\text{Ph}-\text{CH}_2\text{CH}_2\text{OH}$

w) NaOH

x) $\text{Et}-\text{C}(=\text{O})-\text{OCH}_3$ 1) excess EtMgBr
2) H^+ $\rightarrow$ $\text{Et}-\text{C}(\text{OH})(\text{Et})-\text{Et}$

y) $\text{Cyclopentyl bromide}$ 1) Li
2) CuI
3) 1-octyne (excess) $\rightarrow$ $\text{Cyclopentyl octyl ether}$

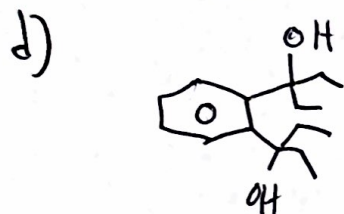
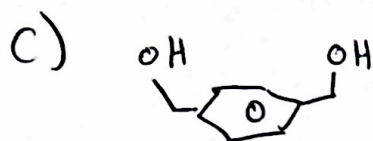
z) OsO_4

aa) $\text{Ph}-\text{Br}$ 1) Mg
2) Cyclohexanone
3) H^+ $\rightarrow$ $\text{Ph}-\text{C}(\text{OH})(\text{H})-\text{Cyclohexyl}$

bb) NaSH

25) a) A meta B ortho C para

b) No Reaction

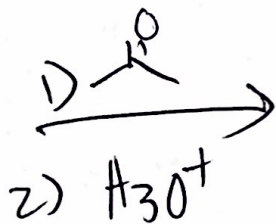


26) a) H on N would quench the Grignard reagent

b) The H on OH would Quench the Grignard Reagent made.

c) The H on the Carbox. Acid would quench the Grignard Reag.

d) without using numbers for the Reagents on the arrow. IT means throwing them in all at the same time. The H_2O^+ would Quench the Grignard Reagent.



is correct.